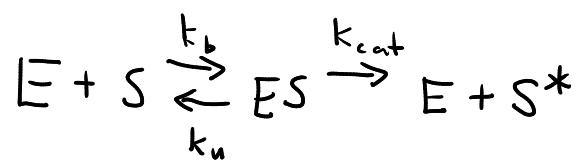


$$\vec{n} = (n_s, n_E, n_{ES}, n_{S^*})$$



$$n_s + n_{ES} + n_{S^*} = M_s$$

$$n_E + n_{ES} = M_E$$

$$M_s = 2, M_E = 2$$

$$n_s \quad n_E \quad n_{ES} \quad n_{S^*}$$

$$\vec{n} = (2 \quad 2 \quad 0 \quad 0)$$

binding

$$(1 \quad 1 \quad 1 \quad 0)$$

catalysis

$$(1 \quad 2 \quad 0 \quad 1)$$

all possible non-zero off-diag. entries of  
 $\Omega_{\vec{n}, \vec{m}}$  for  $\vec{n} \neq \vec{m}$  (all ways  $\vec{m} \rightarrow \vec{n}$ )  
 start end

i) binding

$$\vec{n} = (n_s, n_E, n_{ES}, n_{S^*}) \quad \text{end}$$

$$\vec{m} = (m_s, m_E, m_{ES}, m_{S^*}) \quad \text{start}$$

if

$$n_s = m_s - 1$$

$$n_E = m_E - 1$$

$$n_{ES} = m_{ES} + 1$$

$$n_{S^*} = m_{S^*}$$

is true  $\Rightarrow$

b/c there  
are  $m_E$   
targets

$$\Omega_{\vec{n}, \vec{m}} = \alpha_b K_s \frac{m_s}{V} m_E$$

cell  
volume

binding reaction  $\sim \alpha_b K_s C$   
 (single target)  $\text{conc. of searcher}$   $= \tilde{K}_b m_s m_E$

$$\tilde{K}_b = \frac{\alpha_b K_s}{V}$$

ii) unbinding

if  $n_E = m_E + 1$

$n_s = m_s + 1$

$n_{ES} = m_{ES} - 1$

$n_{S^*} = m_{S^*}$

$$\Omega_{\vec{n}, \vec{m}} = K_u m_{ES}$$

### iii) catalysis

if  $n_E = m_E + 1$   
 $n_S = m_S$  is true  $\Rightarrow \Omega_{\vec{n}, \vec{m}} = k_{cat} m_E s$

$$n_{ES} = m_{ES} - 1$$

$$n_{S*} = m_{S*} + 1$$

portion of  $\Omega$ :

	(2, 2, 0, 0)	(1, 1, 1, 0)	(1, 2, 0, 1)	...
(2, 2, 0, 0)	$\sim$	$k_u$		
(1, 1, 1, 0)	$4\tilde{k}_b$	$\sim$		
(1, 2, 0, 1)	0	$k_{cat}$	$\sim$	
...				

diag:  $\sim$  make sure columns sum to zero

TRICK: recall  $\langle i \rangle_t = \sum_i i p_i(t)$   
 derived  $\frac{d\langle i \rangle_t}{dt} = \sum_{i,j} (j-i) \Omega_{ji} p_i(t)$

general version:  $\langle n_s \rangle_t = \sum_{\vec{n}} n_s p_{\vec{n}}(t)$

$$\langle n_E \rangle_t = \sum_{\vec{n}} n_E p_{\vec{n}}(t)$$

etc.

1)  $\frac{d\langle n_s \rangle_t}{dt} = \sum_{\vec{m}, \vec{n}} (n_s - m_s) \Omega_{\vec{n}, \vec{m}} p_{\vec{m}}(t)$

2)  $\frac{d\langle n_E \rangle_t}{dt} = \sum_{\vec{m}, \vec{n}} (n_E - m_E) \Omega_{\vec{n}, \vec{m}} p_{\vec{m}}(t)$

• • •      4 eqn's

plug in  $\Omega$  : 1)  $\frac{d\langle n_s \rangle_t}{dt} = \sum_m \left[ -\tilde{k}_b m_s m_E \bar{P_m}(+) + k_u m_{ES} \bar{P_m}(+) \right]$

• • •

$\Rightarrow$  4 eqn's:

exact  
eqn's for  
avg's

1)  $\frac{d\langle n_s \rangle_t}{dt} = -\tilde{k}_b \langle n_s n_E \rangle_t + k_u \langle n_{ES} \rangle_t$

2)  $\frac{d\langle n_E \rangle_t}{dt} = -\tilde{k}_b \langle n_s n_E \rangle_t + (k_u + k_{cat}) \cdot \langle n_{ES} \rangle_t$

3)  $\frac{d\langle n_{ES} \rangle_t}{dt} = \tilde{k}_b \langle n_s n_E \rangle_t - (k_u + k_{cat}) \langle n_{ES} \rangle_t$

4)  $\frac{d\langle n_s^* \rangle_t}{dt} = k_{cat} \langle n_{ES} \rangle_t$

5 unknown funcs:  $\langle n_s \rangle_t, \langle n_E \rangle_t, \langle n_{ES} \rangle_t, \langle n_s^* \rangle_t, \langle n_s n_E \rangle_t$

but only 4 equis

"dirty trick":  $\langle n_s n_E \rangle_t \approx \langle n_s \rangle_t \langle n_E \rangle_t$

note: gives 4 eqns + 4 unknowns  $\Rightarrow$   
can solve!

Why is this justified?

digression: imagine two variables describing a system,  $X$  and  $Y$

joint prob. distribution  $P(X, Y)$   
 ↳ frac. of pop. w/ value  $(X, Y)$

marginal prob.  $P(X) = \sum_Y P(X, Y)$

$$P(Y) = \sum_X P(X, Y)$$

$$\langle X \rangle = \sum_{x,y} x P(x, y) = \sum_x x P(x)$$

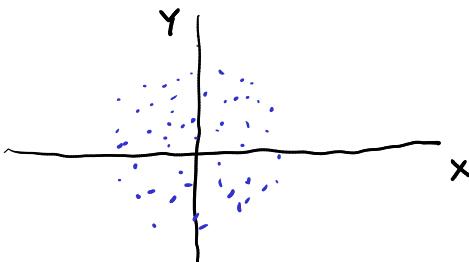
$$\langle Y \rangle = \sum_{x,y} y P(x, y) = \sum_y y P(y)$$

$$\langle XY \rangle = \sum_{x,y} xy P(x, y)$$

in general  $\langle XY \rangle \neq \langle X \rangle \langle Y \rangle$

when is  $\langle XY \rangle \approx \langle X \rangle \langle Y \rangle$

diff. cases:

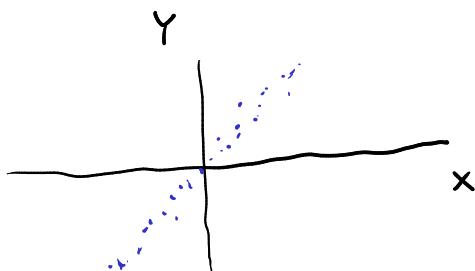


$X \leftrightarrow Y$  not strongly correlated

$$P(X, Y) \approx P(X) P(Y)$$

$$\langle XY \rangle \approx \langle X \rangle \langle Y \rangle$$

opposite case:

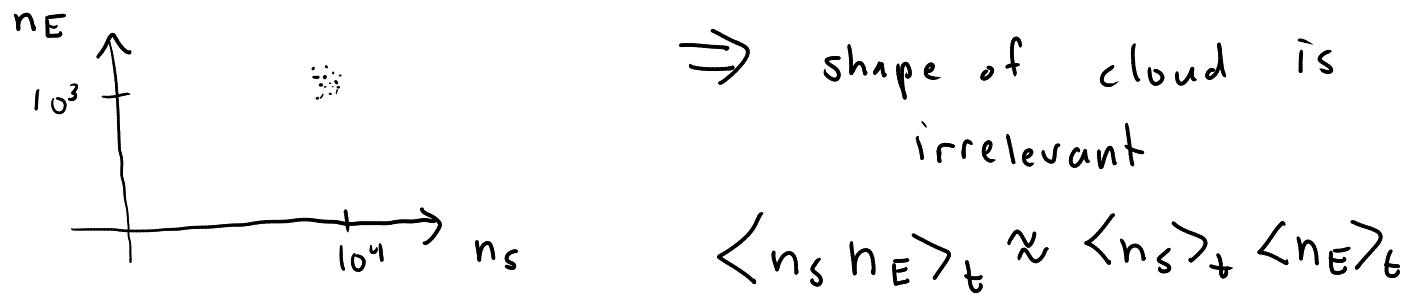


strong correlations

$$\langle XY \rangle \neq \langle X \rangle \langle Y \rangle$$

for chemical systems, mean number of chemical types (# mol's) is quite high

( $10^2$ - $10^5$  mol /cell) + fluctuations  $\ll$  mean #



define concentrations:  $c_s(t) = \frac{\langle n_s \rangle_t}{V}$

$$\frac{dc_s}{dt} = -\tilde{k}_b V c_s c_E + k_u c_{ES}$$

$$\frac{dc_E}{dt} = -\tilde{k}_b V c_s c_E + (k_u + k_{cat}) c_{ES}$$

$$\frac{dc_{ES}}{dt} = \tilde{k}_b V c_s c_E - (k_u + k_{cat}) c_{ES}$$

$$\frac{dc_s^*}{dt} = k_{cat} c_{ES}$$

4 eqn's for  
 $c_E, c_s, c_{ES}, c_s^*$

chemical kinetics

"law of mass action"