

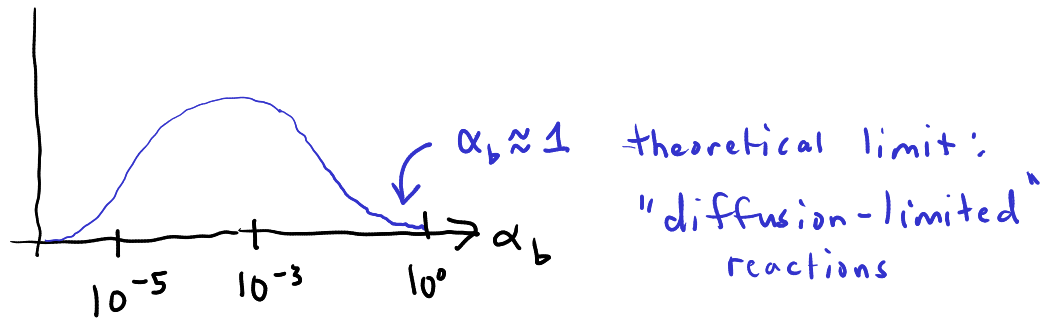
$$0 \leq \alpha_b \leq 1$$

$$\Rightarrow k_b = \alpha_b k_{dif} = \alpha_b k_s c \leq k_{dif}$$

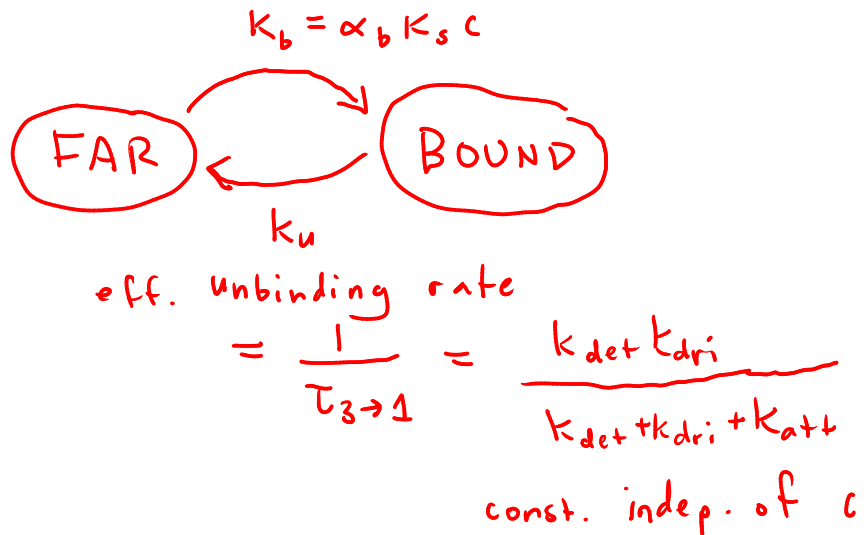
"speed limit"

since $k_{dif} \ll k_{dri} \Rightarrow \alpha_b \approx \frac{k_{att}}{k_{dri} + k_{att}}$ approx. indep. of c

histogram of α_b over all biochemical reactions

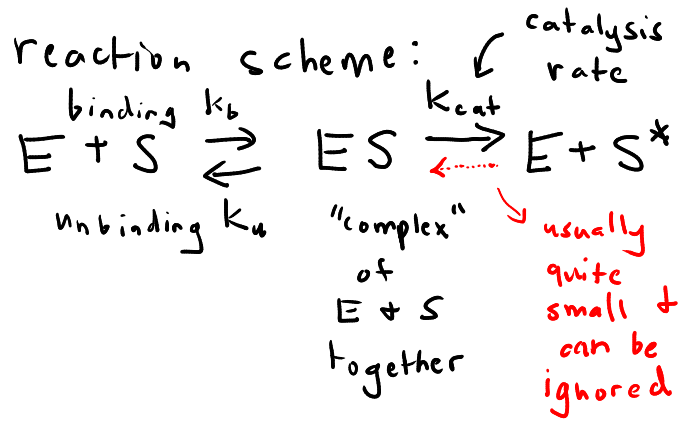
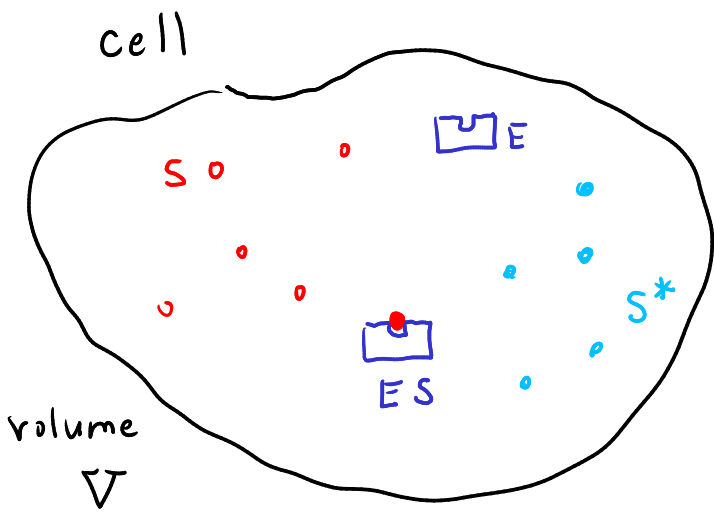


effective 2 state approx.



focus on a full chemical reaction:

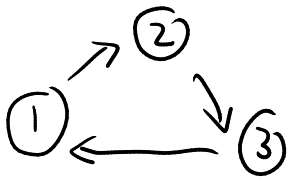
enzyme	E	(one type of protein)
substrate	S	(another protein or small molecule)
reaction	$S \rightarrow S^*$	(modified version of substrate)



chemical state:

$$\vec{n} = (n_S, n_E, n_{ES}, n_{S^*})$$

$n_\alpha = \#$ of type α in cell



Ω_{nm} $m \rightarrow n$ generalize

$\Omega_{\vec{n}\vec{m}} \equiv$ trans. rate from state \vec{m} to \vec{n} (for $\vec{m} \neq \vec{n}$)

probability $P_{\vec{n}}(t) =$ prob. to be in state \vec{n} at time t

$$\sum_{\vec{n}} P_{\vec{n}}(t) = 1$$

$$\sum_{n_S=0}^{\infty} \sum_{n_E=0}^{\infty} \dots$$

columns of $\Omega_{\vec{n}\vec{m}}$ sum to zero

$$\sum_{\vec{n}} \Omega_{\vec{n}\vec{m}} = 0$$

$$\frac{d}{dt} P_{\vec{n}}(t) = \sum_{\vec{m}} \Omega_{\vec{n}\vec{m}} P_{\vec{m}}(t)$$

chemical master equ.

Conservation laws (stoichiometry)

$$n_S + n_{ES} + n_{S^*} = \text{const.} \equiv M_S$$

$$n_E + n_{ES} = \text{const.} \equiv M_E$$

(accurate, but hard way of doing chemistry)

example:

$$M_S = 2,$$

$$M_E = 2$$

allowed
states

$$\vec{v} = \begin{pmatrix} n_s & n_E & n_{ES} & n_{S^*} \\ 2 & 2 & 0 & 0 \\ 1 & 1 & 1 & 0 \\ 1 & 2 & 0 & 1 \\ \dots & \dots & \dots & \dots \\ 0 & 0 & 2 & 0 \\ 0 & 1 & 1 & 1 \end{pmatrix}$$

catalysis \downarrow

binding \downarrow

catalysis \downarrow

binding \downarrow