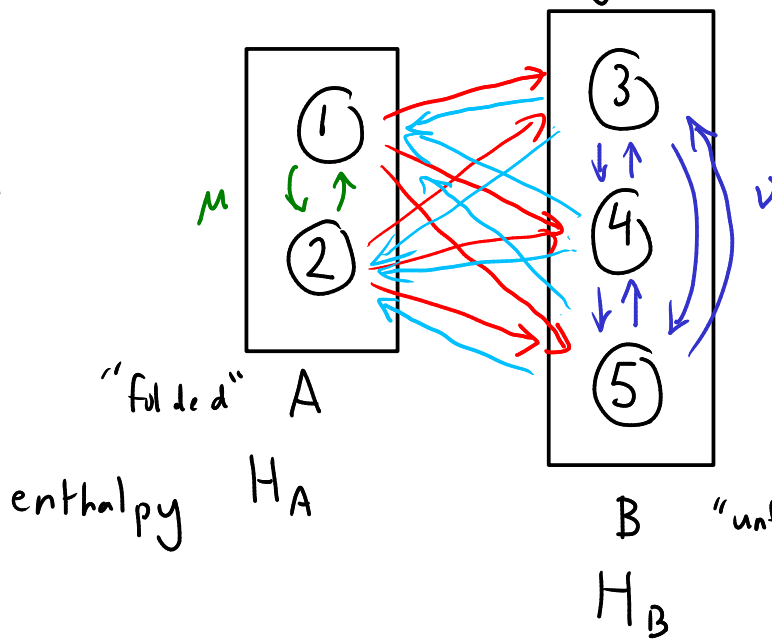


example of coarse-graining:

$$\Omega_A = 2$$



k_u unfolding
 k_f folding

$$\mu, \nu \gg k_u, k_f$$

$$\Omega_B = \# \text{ microstates in } B = 3$$

write down master equ. for microstates:

$$\frac{dp_1}{dt} = -3k_u p_1 - \mu p_1 + k_f (p_3 + p_4 + p_5) + \mu p_2$$

$$\frac{dp_2}{dt} = -3k_u p_2 - \mu p_2 + k_f (p_3 + p_4 + p_5) + \mu p_1$$

...

coarse-grained prob:

$$p_A = p_1 + p_2$$

$$p_B = p_3 + p_4 + p_5$$

$$\Rightarrow \frac{d}{dt} (p_1 + p_2) = -3k_u (p_1 + p_2) + 2k_f (p_3 + p_4 + p_5)$$

$$\Rightarrow \frac{d}{dt} p_A = -3k_u p_A + 2k_f p_B$$

similar steps $\Rightarrow \frac{d}{dt} p_B = 3k_u p_A - 2k_f p_B$

$$\frac{d}{dt} \begin{pmatrix} p_A \\ p_B \end{pmatrix} = \Omega \begin{pmatrix} p_A \\ p_B \end{pmatrix}$$

$$\begin{array}{c} \textcircled{A} \xrightleftharpoons[\Omega_{AB}]{\Omega_{BA}} \textcircled{B} \\ \Omega_{BA} = 3k_u = \delta_B k_u \text{ in general} \\ \Omega_{AB} = 2k_f = \delta_A k_f \text{ in general} \end{array}$$

rate from
microstate
to microstate
↓

generalize: macrostates

$$\begin{array}{c} \textcircled{n} \xrightleftharpoons[\Omega_{nm}]{\Omega_{mn}} \textcircled{m} \\ \Omega_{mn} = \delta_m k_{mn} \\ \Omega_{nm} = \delta_n k_{nm} \end{array}$$

microstates obey LDB:

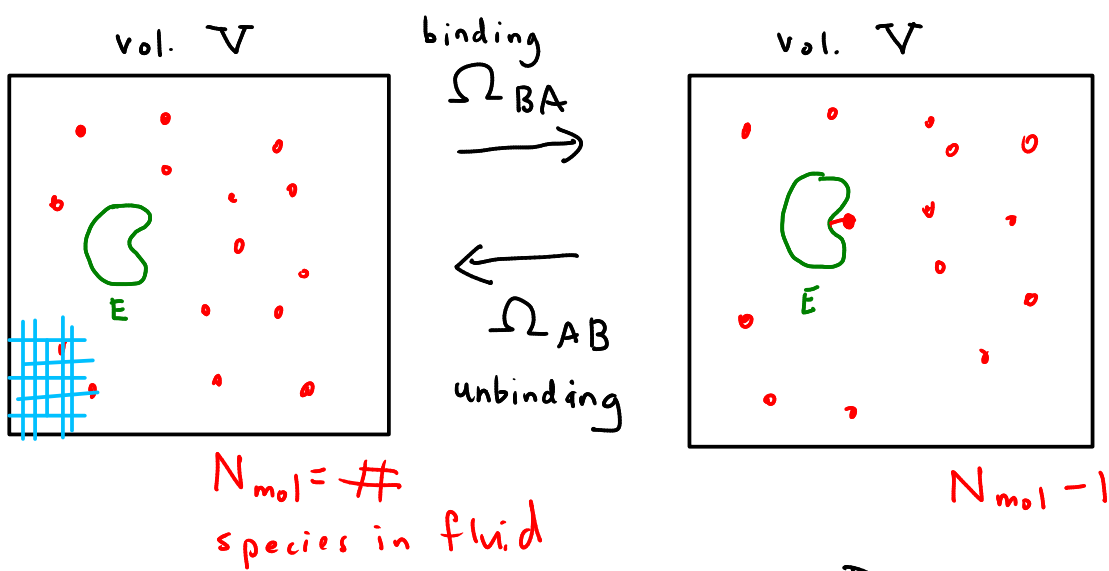
$$\frac{k_{mn}}{k_{nm}} = e^{-\beta(H_m - H_n - W_{mn})}$$

↑ other work

$$\begin{aligned} \Rightarrow \frac{\Omega_{nm}}{\Omega_{mn}} &= \frac{\delta_n k_{nm}}{\delta_m k_{mn}} = \frac{\delta_n}{\delta_m} e^{-\beta(H_n - H_m - W_{nm})} \\ &= e^{-\beta(H_n - H_m - \underbrace{k_B T \ln \frac{\delta_n}{\delta_m}}_{\text{effective "work" term coming from coarse-graining}} - W_{nm})} \end{aligned}$$

example:

enzyme
at
fixed
pos.



macrostate A
= all configs of
 N_{mol} in fluid

B
= all configs of
 $N_{mol} - 1$ in fluid
+ 1 bound to enzyme

binding Ω_{BA}

unbinding Ω_{AB}

$$\frac{\Omega_{BA}}{\Omega_{AB}} = e^{-\beta \left(\underbrace{H_B - H_A}_{< 0 \text{ for molecule forming a bond}} - k_B T \ln \frac{\Omega_B}{\Omega_A} \right)}$$

(if binding is thermod. favorable)

$W_{BA} = 0$
no other work terms

$\Omega_A = \#$ configs of N_{mol} molecules in
a box of volume V (ignore vol. of E
 $\ll V$)

divide up vol. V into small volumes

$V_{mol} = \text{vol. of molecule (big enough to fit one molec.)}$

$$\# \text{ boxes} = N_{\text{pos}} = \frac{V}{V_{\text{mol}}}$$

G_A = how many ways to arrange N_{mol} objects into N_{pos} possible positions?

$$G_A = \binom{N_{\text{pos}}}{N_{\text{mol}}} = \frac{N_{\text{pos}}!}{N_{\text{mol}}! (N_{\text{pos}} - N_{\text{mol}})!}$$

$$G_B = \binom{N_{\text{pos}}}{N_{\text{mol}} - 1} = \frac{N_{\text{pos}}!}{(N_{\text{mol}} - 1)! (N_{\text{pos}} - N_{\text{mol}} + 1)!}$$

$$\frac{G_B}{G_A} = \frac{N_{\text{mol}}}{N_{\text{pos}} - N_{\text{mol}} + 1} \approx \frac{N_{\text{mol}}}{N_{\text{pos}}} \quad \begin{array}{l} N_{\text{pos}} \gg N_{\text{mol}} \\ \text{(dilute sol'n)} \end{array}$$

$$\begin{aligned} k_B T \ln \frac{G_B}{G_A} &\approx k_B T \ln \frac{N_{\text{mol}}}{N_{\text{pos}}} \\ &= k_B T \ln c V_{\text{mol}} \\ &= \underbrace{k_B T \ln V_{\text{mol}}}_{\mu_0 = \text{const.}} + k_B T \ln c \\ &= \text{"standard"} \\ &\quad \text{chem. potential} \end{aligned}$$

$$N_{\text{pos}} = \frac{V}{V_{\text{mol}}}$$

$$c = \frac{N_{\text{mol}}}{V} = \text{conc.}$$

$$\text{chemical potential} \equiv \mu = k_B T \ln \frac{G_B}{G_A}$$

$$= \mu_0 + k_B T \ln c$$

$$\frac{\text{binding}}{\text{unbinding}} \quad \frac{\Omega_{BA}}{\Omega_{AB}} = e^{-\beta (\underbrace{H_B - H_A}_{\substack{< 0 \\ \text{makes} \\ \text{binding} \\ \text{favorable}}} - \mu)} \quad \text{Eq. 1}$$

$$\text{When } N_{\text{pos}} \gg N_{\text{mol}} \Rightarrow \mu < 0$$

$$\text{Increasing } c \Rightarrow \mu \text{ less negative}$$

$$\Rightarrow \text{makes } \Omega_{BA} \text{ larger rel. to } \Omega_{AB}$$

$$\Rightarrow \text{larger conc. } c \text{ favors binding}$$

recall from

$$\text{FAR} \xleftrightarrow{\text{NEAR}} \text{BOUND}$$

$0 \leq \alpha_b \leq 1$
Smol. rate

$$\text{Eq. 2} \quad \frac{k_b}{k_u} = \frac{\Omega_{BA}}{\Omega_{AB}} = \frac{\alpha_b K_s c}{k_u}$$

both Eq. 1 & Eq. 2 are prop. to c
 $\Rightarrow$ consistent!

next time: generalize to many types
of particles in fluid around
our system

type j molec: $\mu_j = \mu_o^j + k_B T \ln c_j$

$\hookrightarrow$ conc.
of type
 j molec.