

$$\dot{Q}_1 = \frac{Q_1}{\Delta t}$$

$$\dot{Q}_2 = \frac{Q_2}{\Delta t}$$

$$\delta = \frac{I}{\Delta t} \geq 0$$

$$\dot{Q}_\alpha = \frac{Q_\alpha}{\Delta t} \quad \alpha = 1, 2$$

$$P = \frac{W}{\Delta t}$$

$$\Delta t = \begin{cases} \delta t & \text{stat. state} \\ \tau \delta t & \text{periodic driving} \end{cases}$$

Case 1: $\dot{Q}_2 > 0 \Rightarrow \frac{P}{\dot{Q}_2} = \eta = 1 - \frac{T_1}{T_2} - T_1 \delta$

heat engine:

draws heat from hot bath, dumps into cold bath ($\dot{Q}_1 < 0$) while doing some work ($\bar{W} > 0$)

$$\leq 1 - \frac{T_1}{T_2} \equiv \eta_{\max} \text{ Carnot bound}$$

max. efficiency $\eta \rightarrow \eta_{\max}$ requires $\delta = \frac{I}{\Delta t} \rightarrow 0$

two options:

• $I \rightarrow 0$ (equilibrium)

set $T_1 = T_2$ + let sys. equil. at one temp.

$$\Rightarrow \eta = \eta_{\max} = 1 - \frac{T_1}{T_2} = 0$$

• $\Delta t \rightarrow \infty$ (infinitely long period of driving)

$$I \neq 0 \Rightarrow \bar{W} = Q_1 + Q_2 > 0$$

power $\frac{\bar{W}}{\Delta t} = P \rightarrow 0$ max. efficiency but zero power

more interesting question: what is efficiency at max. power \Rightarrow no universal answer

case 2: $\dot{Q}_1 > 0$ (draws heat from cold bath)

1st: $P = \dot{Q}_1 + \dot{Q}_2$ $\rightarrow \frac{\dot{Q}_2}{T_2} = -\frac{\dot{Q}_1}{T_1} - \delta$

2nd: $\delta = -\frac{\dot{Q}_1}{T_1} - \frac{\dot{Q}_2}{T_2} \geq 0 \Rightarrow \dot{Q}_2 < 0$ (dumps heat into hot bath)

example: refrigerator
heat pump

$$\dot{Q}_2 = -\frac{T_2}{T_1} \dot{Q}_1 - T_2 \delta \quad \frac{T_2}{T_1} > 1$$

$$\Rightarrow |\dot{Q}_2| > |\dot{Q}_1|$$

$$\Rightarrow P = \dot{Q}_1 + \dot{Q}_2 = \dot{Q}_1 - |\dot{Q}_2| < 0$$

\Rightarrow plug in your fridge

coeff. of performance $\eta_R = \frac{\dot{Q}_1}{-P} = \frac{\text{heat rate out cold bath}}{\text{input power}}$

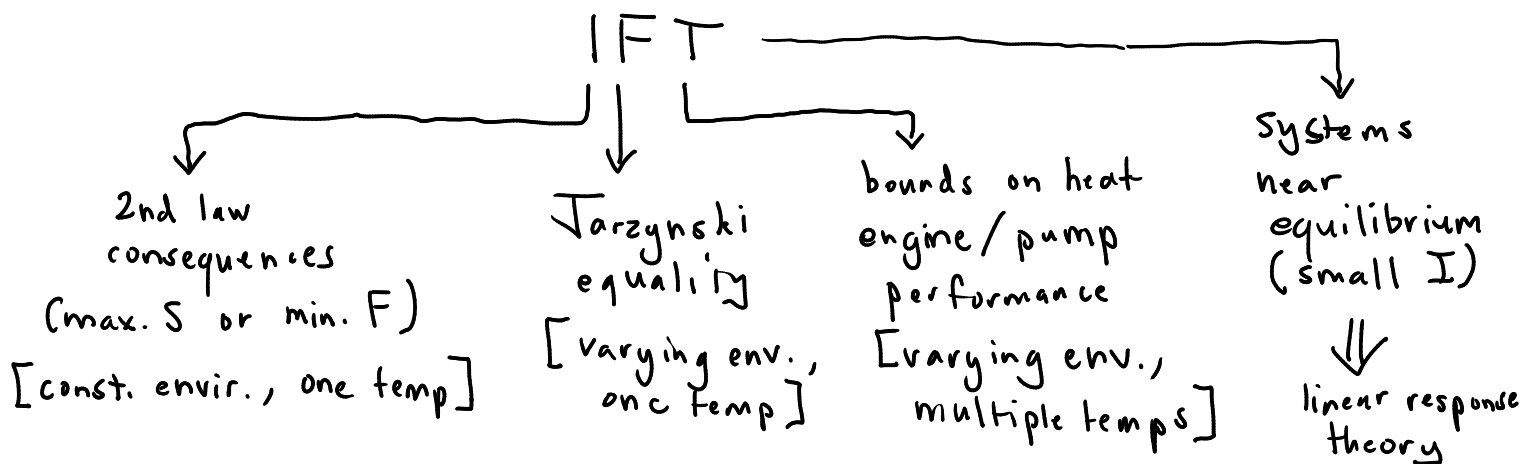
$$\eta_R = \frac{\dot{Q}_1}{-\dot{Q}_1 - \dot{Q}_2} = \text{algebra} = \frac{T_1}{T_2 - T_1} - \frac{T_1 T_2 \delta}{T_2 - T_1} \leq 0$$

$$\eta_R \leq \frac{T_1}{T_2 - T_1} = \eta_R^{\max}$$

$\sim 300 \text{ K}$ $\sim 30 \text{ K}$

η_R^{\max} can be quite large
 $\sim O(10)$

Big picture perspective: "many children of IFT"



Linear thermodynamics

Single temp. T + stationary state (not necessarily equil.)

2nd: $I = -\frac{1}{T} \Delta \dot{F} - \dot{W} \geq 0$

1st: $Q = \Delta \dot{E} + \dot{W}$

$\Rightarrow I = -\frac{1}{T} \dot{W} = -\frac{1}{T} Q \Rightarrow Q = \dot{W} = -T I \leq 0$

heat into env. work on sys.

$I = 0 \Rightarrow$ ESS ($Q = \dot{W} = 0$)

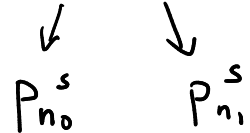
$I > 0 \Rightarrow$ NESS ($Q = \dot{W} < 0$)

'near' equil: I small ($Q = \dot{W}$ also small)

focus on one time step δt :

stationary state: \vec{p}^s

$\mu_0 = (n_0, n_1)$



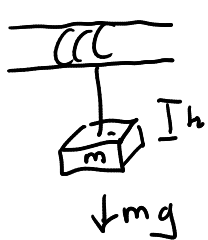
$I = \langle I(\mu_0) \rangle = -\frac{1}{T} \langle \dot{W}(\mu_0) \rangle = -\frac{1}{T} \bar{\dot{W}}$

assume form for work: $\bar{\dot{W}}(\mu_0) = \omega_{n_1, n_0}$

$= -f \Delta x_{n_1, n_0}$

"force" "distance":

(some physical quantity we control) change in a phys. variable



$f = mg$

$\Delta x_{n_1, n_0} = \begin{cases} +h & \text{if } n_1 = n_0 + 1 \\ -h & \text{if } n_1 = n_0 - 1 \\ 0 & \text{otherwise} \end{cases}$